

Properties and units

2.23

A steel cylinder of mass 2 kg contains 4 L of liquid water at 25°C at 200 kPa. Find the total mass and volume of the system. List two extensive and three intensive properties of the water

Solution:

Density of steel in Table A.3: $\rho = 7820 \text{ kg/m}^3$

Volume of steel: $V = m/\rho = \frac{2 \text{ kg}}{7820 \text{ kg/m}^3} = 0.000 256 \text{ m}^3$

Density of water in Table A.4: $\rho = 997 \text{ kg/m}^3$

Mass of water: $m = \rho V = 997 \text{ kg/m}^3 \times 0.004 \text{ m}^3 = 3.988 \text{ kg}$

Total mass: $m = m_{\text{steel}} + m_{\text{water}} = 2 + 3.988 = \mathbf{5.988 \text{ kg}}$

Total volume: $V = V_{\text{steel}} + V_{\text{water}} = 0.000 256 + 0.004$
 $= \mathbf{0.004 256 \text{ m}^3} = \mathbf{4.26 \text{ L}}$

Specific Volume

2.38

A 5 m³ container is filled with 900 kg of granite (density 2400 kg/m³) and the rest of the volume is air with density 1.15 kg/m³. Find the mass of air and the overall (average) specific volume.

Solution:

$$\begin{aligned} m_{\text{air}} &= \rho V = \rho_{\text{air}} \left(V_{\text{tot}} - \frac{m_{\text{granite}}}{\rho} \right) \\ &= 1.15 \left[5 - \frac{900}{2400} \right] = 1.15 \times 4.625 = \mathbf{5.32 \text{ kg}} \\ v &= \frac{V}{m} = \frac{5}{900 + 5.32} = \mathbf{0.00552 \text{ m}^3/\text{kg}} \end{aligned}$$

Comment: Because the air and the granite are not mixed or evenly distributed in the container the overall specific volume or density does not have much meaning.

2.45

A piston/cylinder with cross sectional area of 0.01 m^2 has a piston mass of 100 kg resting on the stops, as shown in Fig. P2.45. With an outside atmospheric pressure of 100 kPa , what should the water pressure be to lift the piston?

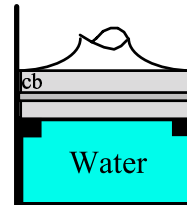
Solution:

The force acting down on the piston comes from gravitation and the outside atmospheric pressure acting over the top surface.

$$\text{Force balance:} \quad F\uparrow = F\downarrow = PA = m_p g + P_0 A$$

Now solve for P (divide by 1000 to convert to kPa for 2nd term)

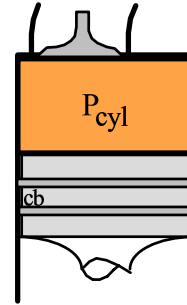
$$\begin{aligned} P &= P_0 + \frac{m_p g}{A} = 100 \text{ kPa} + \frac{100 \times 9.80665}{0.01 \times 1000} \\ &= 100 \text{ kPa} + 98.07 \text{ kPa} = \mathbf{198 \text{ kPa}} \end{aligned}$$



2.47

A valve in a cylinder has a cross sectional area of 11 cm^2 with a pressure of 735 kPa inside the cylinder and 99 kPa outside. How large a force is needed to open the valve?

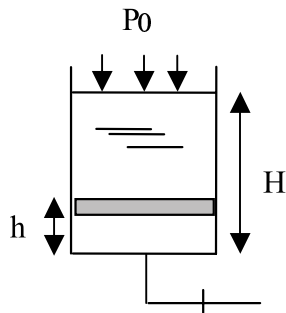
$$\begin{aligned} F_{\text{net}} &= P_{\text{in}}A - P_{\text{out}}A \\ &= (735 - 99) \text{ kPa} \times 11 \text{ cm}^2 \\ &= 6996 \text{ kPa cm}^2 \\ &= 6996 \times \frac{\text{kN}}{\text{m}^2} \times 10^{-4} \text{ m}^2 \\ &= \mathbf{700 \text{ N}} \end{aligned}$$



2.57

Liquid water with density ρ is filled on top of a thin piston in a cylinder with cross-sectional area A and total height H . Air is let in under the piston so it pushes up, spilling the water over the edge. Deduce the formula for the air pressure as a function of the piston elevation from the bottom, h .

Solution:

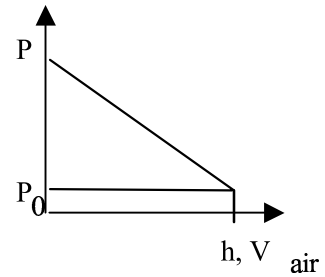


Force balance
Piston: $F\uparrow = F\downarrow$

$$PA = P_0A + m_{\text{H}_2\text{O}}g$$

$$P = P_0 + m_{\text{H}_2\text{O}}g/A$$

$$P = P_0 + (H - h)\rho g$$



2.68

An exploration submarine should be able to go 4000 m down in the ocean. If the ocean density is 1020 kg/m^3 what is the maximum pressure on the submarine hull?

Solution:

$$\begin{aligned}\Delta P &= \rho Lg = (1020 \text{ kg/m}^3 \times 4000 \text{ m} \times 9.807 \text{ m/s}^2) / 1000 \\ &= 40\,012 \text{ kPa} \approx \mathbf{40 \text{ MPa}}\end{aligned}$$

2.82

The main waterline into a tall building has a pressure of 600 kPa at 5 m elevation below ground level. How much extra pressure does a pump need to add to ensure a water line pressure of 200 kPa at the top floor 150 m above ground?

Solution:

The pump exit pressure must balance the top pressure plus the column ΔP . The pump inlet pressure provides part of the absolute pressure.

$$P_{\text{after pump}} = P_{\text{top}} + \Delta P$$

$$\begin{aligned}\Delta P &= \rho gh = 997 \text{ kg/m}^3 \times 9.807 \text{ m/s}^2 \times (150 + 5) \text{ m} \\ &= 1\,515\,525 \text{ Pa} = 1516 \text{ kPa}\end{aligned}$$

$$P_{\text{after pump}} = 200 + 1516 = 1716 \text{ kPa}$$

$$\Delta P_{\text{pump}} = 1716 - 600 = \mathbf{1116 \text{ kPa}}$$

2.84

In the city water tower, water is pumped up to a level 25 m above ground in a pressurized tank with air at 125 kPa over the water surface. This is illustrated in Fig. P2.84. Assuming the water density is 1000 kg/m^3 and standard gravity, find the pressure required to pump more water in at ground level.

Solution:

$$\begin{aligned}\Delta P &= \rho L g \\ &= 1000 \text{ kg/m}^3 \times 25 \text{ m} \times 9.807 \text{ m/s}^2 \\ &= 245\,175 \text{ Pa} = 245.2 \text{ kPa} \\ P_{\text{bottom}} &= P_{\text{top}} + \Delta P \\ &= 125 + 245.2 \\ &= \mathbf{370 \text{ kPa}}\end{aligned}$$

